Page 1411. Column 1, formula IIIa should be the pyrrolidine enamine rather than the cyclopentylamine structure shown.

Alex Nickon, Norman Schwartz, Joseph B. DiGiorgio, and David A. Widdowson: Reactivity and Geometry in Allylic Systems. IV. Stereochemical Factors in the Photosensitized Oxygenation of 5α - and 5β -Cholest-3-enes.

Pagel 712. Column 2, in formulas illustrated at top, numbers 5 and 6 should be interchanged.

Page 1713. At the end of footnote 24, the reference should be to footnote 2.

Page 1716. Column 1, paragraph B, line 18, "enriched in $3\alpha,4\beta$ -dibromo- 5α -cholestane," should read "enriched in $3\alpha,4\beta$ dibromo-5β-cholestane."

Page 1716. Column 1, paragraph entitled "Regeneration of Olefins from Dibromides. A," line 9, "After being dried over sodium and evaporated," should read "After being dried over sodium sulfate and evaporated."

Andre Rosowsky and Edward J. Modest: 2,2,4-Trimethyl-1,2-dihydroquinolines. Preparation and Nuclear Magnetic Resonance Studies.

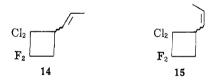
Page 1833. First column, VIII and IX under structure at the bottom of column should read "VIII, $R_1 = R_3 = H$; $R_2 = CH_3$," and "IX, $R_1 = H$; $R_2 = EtO$; $R_3 = CH_3$."

Edward J. Modest, Suprabhat Chatterjee, and Heljo Kangur Protopapa: 2,4-Diaminopyrimidines from Dicyandiamide. III. Reaction with Monocyclic Ketones.

Page 1838. First column, second structure. Legend for IV should read

Nicholas J. Turro and Paul D. Bartlett: Photosensitized Cycloaddition of Haloethylenes and 1,3-Dienes.

Page 1850. Column 2, formulas should read



Page 1851. Column 1, sixth line from bottom, "m/e 80" should read "m/e 66."

Page 1851. Column 2, paragraph 4, line 2, "C7H2CL4" should read "C7H6Cl4."

Page 1851. Column 2, paragraph 4, line 2, "m/e 76" should read "m/e 66."

F. E. Condon and Andreas A. Zavitsas: Synthesis and Isomerization of 2,6-Dimethyl-n-butylbenzene.

Page 1901. Column 1, line 6, "3,5-dimethyl-t-butylbenzene (III)" should read "3,5-dimethyl-sec-butylbenzene."

Timothy F. Parsons, John D. Buckman, D. E. Pearson, and Lamar Field: Organic Disulfides and Related Substances. XIV. Aspects of the Reaction of Thiolsulfonates with Thiols. Page 1926. Column 1, line 24, "25. ml." should read "2.5 ml."

S. Carlton Dickerman, Derek DeSouza, and Philip Wolf: Synthesis and Spectra of Di- and Polyphenylanthracenes.

Page 1981. Line 5 of the abstract, add "or 95% ethanol" after "methylene chloride."

Page 1983. Column 2, line 21, delete "all determined in methylene chloride."

Page 1983. In Table II, the ultraviolet absorption data for 9,10-, 1,4-, 1,9-, 1,10-, 2,9-, and 2,10-diphenylanthracenes were measured in 95% ethanol, not methylene chloride as stated. The shifts in the longest wave length band should read 15, 14, 14, 15, 18, and 16 m μ , respectively. Footnote a should read "Solvent: methylene chloride or 95% ethanol." Add to footnote b "in the same solvent."

John R. Carson, George I. Poos, and Harold R. Almond, Jr.: 2-Amino-5-aryl-2-oxazolines. Tautomerism, Stereochemistry, and an Unusual Reaction.

Page 2226. The structural formulas of VI and VII should

W. D. Crow and Nelson J. Leonard: 3-Isothiazolone-cis-3-Thiocyanoacrylamide Equilibria.

Page 2664. Column 2, line 20, molecular formula should read $C_4H_4N_2OS$.

Francis T. Williams, Jr., Pat W. K. Flanagan, William J. Taylor, and Harold Shechter: Ultraviolet Spectra of Anions of Mononitro Compounds.

Page 2674. Reinvestigation of the ultraviolet absorption of alkanenitronate anions below 220 mu on a nitrogen-purged Beckman DK-2A reveals that the end absorption indicated in Figure 5 is due to sodium hydroxide and not the nitronate. The figure should be disregarded.

The absorption of sodium ethanenitronate centered at 229 mu falls smoothly to about 205 m μ but more gradually than on the long wave length side. However, the $\psi_2 \rightarrow \psi_4$ transition predicted from the calculation is not confirmed (or refuted).

Measurement of the absorption of the ethanenitronate below 210 m μ becomes increasingly unreliable because the end absorption of 0.001 N sodium hydroxide is so intense. Reducing the sodium hydroxide concentration significantly raises the possibility of incomplete conversion of nitroethane (pK = 8.5) to its anion, and nitroethane itself exhibits strong end absorption $[\lambda_{max}]$ $\sim 202 \text{ m}\mu \ (\epsilon_{202} > 5000)$].

John E. Gordon: Fused Organic Salts. III. Chemical Stability of Molten Tetra-n-alkylammonium Salts. Medium Effects on Thermal $R_4N^+X^-$ Decomposition. $RBr + I^- =$ RI + Br - Equilibrium Constant in Fused Salt Medium.

Page 2761. Column 1, line 11, "401°" should read "140°." James W. Patton and Marion O. Son. The Synthesis of Naphthalene-2,3-dicarboxylic Acid by the Henkel Process. Page 2869. Column 2, conclusion 3, "The pressure of iron

oxides" should read "The presence of iron oxides."

C. G. Overberger, H. Ringsdorf, and B. Avchen: Potential Antiradiation Agents. Preparation and Polymerization of S-Vinyl-N-vinylthiocarbamates.

Page 3088. Footnote 3 should read "This is the XXXth in a series of papers. . .'

Ernest L. Eliel and Jyotirmoy Roy: Reductions with Metal Hydrides. XVII. Reduction of 1,3-Thiazanes.

Page 3092. Column 2, line 13, " $(\tau 3.2)$ " should read 2.1).

Page 3094. Column 1, line 48, " $(\tau 3.21)$ " should read " $(\tau$ 2.16).

Norman Rabjohn and M. C. Chacon: The Reaction of Lead Tetraacetate with Carbethoxyhydrazones.

Page 3227. The bottom line of structures, column 2, should read

$$(C_6H_5)_2\overset{\text{Br}}{\overset{}{\overset{}{\bigcirc}}} C-C-CH_3 \xrightarrow{H_2O} (C_6H_5)_2-C-C-CH_3$$